



Received:

2026/07/26

Accepted:

2026/09/01

Published:

2026/09/06



REVIEW

OPEN ACCESS

Nanomaterials and their Applications in Energy and Environment: A Comprehensive Review

V Geetha

Lecturer in Chemistry, GDC, Andhra Pradesh, India

*Correspondence for materials should be addressed to VG (email: velalamgeetha@gmail.com)

Abstract

Nanomaterials have emerged as promising materials for addressing major challenges in energy production, storage, environmental protection, and sustainable development. Their unique physicochemical properties, including high surface area, tunable optical, electrical, thermal, and catalytic characteristics, enable improved performance in a wide range of applications. This review highlights recent advances in nanomaterials for energy conversion and storage, including solar cells, batteries, supercapacitors, fuel cells, and hydrogen production. It also examines their environmental applications in wastewater treatment, air-pollution control, pollutant degradation, environmental sensing, and carbon capture. The review discusses the advantages, limitations, environmental risks, and future prospects of nanomaterial-based technologies, emphasizing their potential contribution to clean energy and sustainable environmental management.

Keywords: Nanomaterials; Energy Storage; Energy Conversion; Environmental Remediation; Nanocatalysts; Sustainable Development

Introduction

The rapid growth of the global population, industrialization, and urbanization has significantly increased the demand for energy while simultaneously intensifying environmental degradation. Conventional fossil fuels continue to dominate the world's energy supply, but their excessive consumption has resulted in severe environmental consequences, including greenhouse gas emissions, climate change, air pollution, and depletion of natural resources. Consequently, the development of clean energy technologies and sustainable environmental management strategies has become one of the foremost priorities for researchers, policymakers, and industries worldwide. Nanotechnology has emerged as a transformative field capable of addressing many of these global challenges. By manipulating materials at the nanoscale (1–100 nm), scientists have developed nanomaterials with extraordinary physicochemical properties that are not observed in bulk materials. Their exceptionally high surface area-to-volume ratio, tunable electronic structures, enhanced catalytic activity, superior mechanical strength, and improved electrical and thermal conductivity make them highly attractive for a wide range of scientific and technological applications. Nanomaterials can be broadly classified into metallic nanoparticles, metal oxide nanoparticles, carbon-based nanomaterials, polymeric nanoparticles, semiconductor quantum dots, and hybrid nanocomposites. Each class exhibits unique structural and functional characteristics that enable specific applications in energy conversion, energy storage, catalysis, environmental remediation, biomedical engineering, and electronic devices. Advances in synthesis techniques, including chemical, physical, biological, and green synthesis methods, have further accelerated the development of highly efficient and environmentally friendly nanomaterials. One of the most significant applications of nanomaterials is in the energy sector. They have substantially improved the performance of renewable energy technologies such as photovoltaic cells, fuel cells, hydrogen production systems, lithium-ion batteries, sodium-ion batteries, and supercapacitors. Nanostructured electrodes and catalysts enhance charge transport, increase energy conversion efficiency, improve storage capacity, and extend the operational lifetime of these devices.



As a result, nanotechnology is playing an increasingly important role in facilitating the global transition toward sustainable and low-carbon energy systems. Nanomaterials have also transformed environmental protection technologies. Their remarkable adsorption capacity, photocatalytic efficiency, antimicrobial activity, and chemical stability have enabled effective removal of pollutants from water, air, and soil. Nanomaterial-based filtration membranes, photocatalysts, Nano sorbents, and environmental sensors have demonstrated excellent performance in removing heavy metals, organic contaminants, pathogens, pharmaceutical residues, and emerging pollutants.

These technologies contribute significantly to improving environmental quality and public health. Despite these promising developments, several challenges continue to limit the large-scale implementation of nanomaterials. Potential concerns include nanotoxicity, environmental persistence, safe disposal, production costs, scalability, regulatory compliance, and public acceptance. Addressing these issues requires multidisciplinary collaboration among materials scientists, environmental engineers, chemists, policymakers, and regulatory agencies. Furthermore, sustainable manufacturing approaches, green synthesis techniques, life-cycle assessments, and comprehensive risk evaluations are essential for ensuring the safe and responsible use of nanotechnology. The objective of this review is to provide a comprehensive overview of the latest developments in nanomaterials for energy and environmental applications. The paper examines the different classes of nanomaterials, their synthesis methods, and their physicochemical properties before discussing their diverse applications in renewable energy generation, energy storage, pollution control, water treatment, air purification, and environmental monitoring. It also highlights current challenges, recent innovations, and future research directions, emphasizing the crucial role of nanotechnology in achieving global sustainability and supporting the transition toward a cleaner and more energy-efficient future.

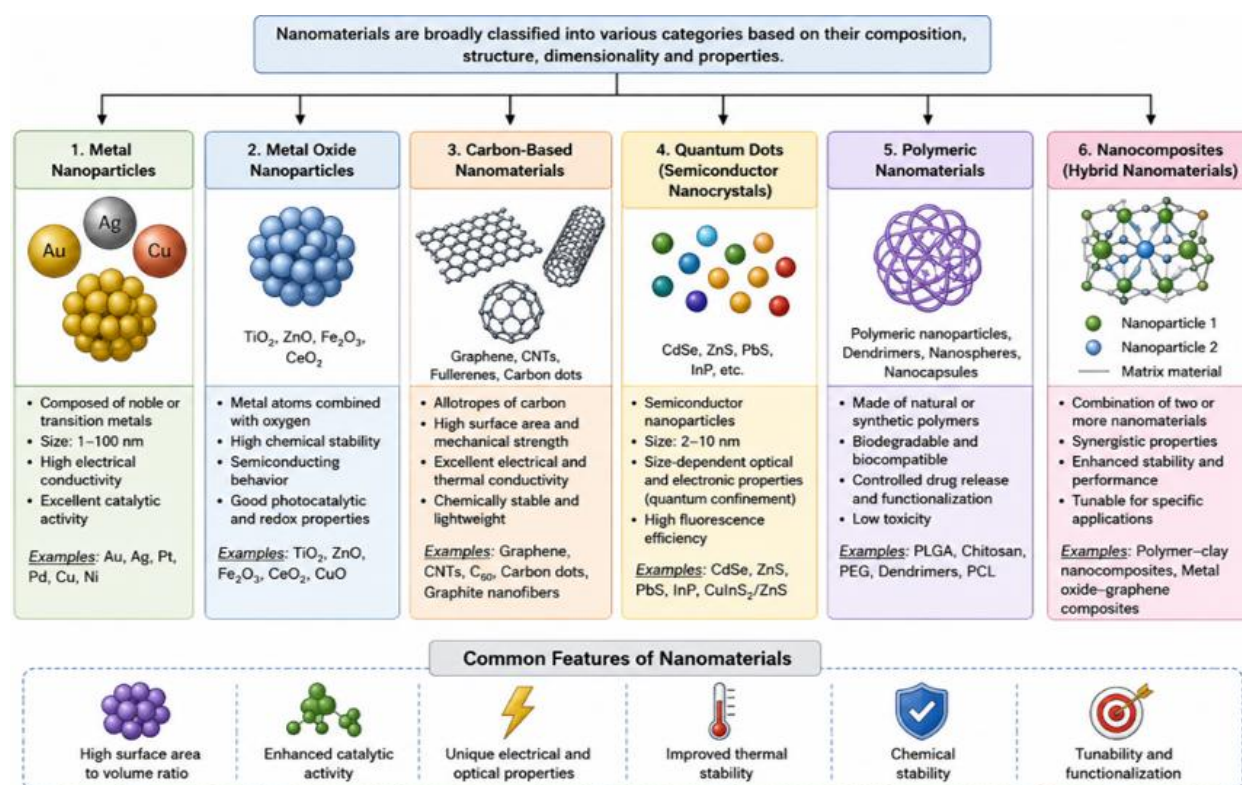


Fig. 1. Classification of Nanomaterials

Applications of Nanomaterials in Energy

The increasing demand for clean, efficient, and sustainable energy has accelerated the development of advanced materials capable of improving energy generation, conversion, and storage technologies. Conventional energy systems are often limited by low efficiency, resource depletion, and environmental pollution. Nanomaterials have emerged as promising alternatives because of their high surface area, unique electronic properties, excellent catalytic activity, and enhanced charge transport characteristics. These properties significantly improve the performance of renewable energy devices and contribute to the global transition toward carbon-neutral energy systems.

Nanomaterials in Solar Energy Conversion

Solar energy is one of the most abundant renewable energy sources available. Nanomaterials have significantly enhanced photovoltaic technologies by improving light absorption, charge separation, and electron transport. Metal oxide nanoparticles such as titanium dioxide (TiO₂) and zinc oxide (ZnO) are widely used in dye-sensitized

and perovskite solar cells because of their high electron mobility and excellent stability. Carbon nanomaterials, including graphene and carbon nanotubes (CNTs), improve electrical conductivity and reduce charge recombination losses. Quantum dots enable the absorption of a broader range of the solar spectrum, increasing overall conversion efficiency.

Nanomaterials in Lithium-Ion Batteries

Rechargeable lithium-ion batteries power electric vehicles, portable electronics, and renewable energy storage systems. Nanostructured electrode materials improve battery performance by shortening lithium-ion diffusion pathways and increasing electrode surface area. Nanomaterials such as silicon nanoparticles, lithium iron phosphate (LiFePO₄), graphene, titanium dioxide, and carbon nanotubes enhance battery capacity, charging speed, cycling stability, and energy density. Nanomaterials in Supercapacitors: Supercapacitors bridge the gap between conventional capacitors and batteries by providing rapid energy storage and discharge. Nanomaterials increase electrode surface area, allowing greater charge accumulation. Graphene, activated carbon, carbon nanotubes, metal oxides (MnO₂, RuO₂), and conductive polymers are widely employed as electrode materials.

Nanomaterials in Fuel Cells

Fuel cells convert chemical energy directly into electrical energy with high efficiency and minimal environmental impact. Nanocatalysts reduce reaction barriers and improve electrochemical performance. Platinum nanoparticles remain the most efficient catalysts for hydrogen fuel cells, while graphene-supported catalysts and transition metal nanomaterials are being developed to reduce costs and improve durability. Applications include electric vehicles, Portable power systems, Backup power generation, Aerospace technologies. Advantages-High efficiency, Low greenhouse gas emissions, Improved catalyst utilization.

Nanomaterials in Hydrogen Production and Storage

Hydrogen is considered a clean energy carrier capable of replacing fossil fuels in many applications. Nanomaterials improve hydrogen production through photocatalytic water splitting and enhance hydrogen storage capacity. Titanium dioxide nanoparticles, graphitic carbon nitride (g-C₃N₄), molybdenum disulfide (MoS₂), and metal-organic frameworks (MOFs) have demonstrated excellent catalytic performance for hydrogen evolution. Nanostructured metal hydrides, carbon nanotubes, graphene, and porous nanomaterials provide efficient hydrogen storage by increasing adsorption capacity.

Nanomaterials in Thermoelectric Energy Conversion

Thermoelectric materials convert waste heat directly into electricity. Nano structuring reduces thermal conductivity while maintaining electrical conductivity, thereby increasing energy conversion efficiency. Common nanomaterials include: Bismuth telluride (Bi₂Te₃), Lead telluride (PbTe), Silicon nanowires, Graphene composites. Applications-Industrial waste heat recovery, Automotive exhaust systems, Wearable electronics, Space power systems.

Nanocatalysts for Clean Energy Production

Nanocatalysts accelerate chemical reactions while reducing energy consumption and catalyst loading. Their high surface area exposes more active sites, leading to greater catalytic efficiency. Applications include: Water splitting, Carbon dioxide reduction, Biomass conversion, Biofuel production, Artificial photosynthesis.

Common Nanocatalysts

Platinum nanoparticles, Palladium nanoparticles, Nickel nanoparticles, Iron oxide nanoparticles, Titanium dioxide nanoparticles. The incorporation of nanomaterials into energy technologies provides several significant advantages: Large surface area for enhanced reaction rates. Improved electrical and thermal conductivity. Higher energy conversion efficiency. Faster charge transport. Reduced material consumption. Increased device lifetime. Lightweight and compact designs. Better mechanical stability. Enhanced catalytic performance. Support for sustainable and renewable energy technologies.

Challenges

Despite remarkable progress, several challenges remain before widespread commercialization can be achieved: High manufacturing costs, Large-scale synthesis difficulties, Long-term material stability, Environmental and health concerns related to nanoparticle exposure, Recycling and disposal issues, Dependence on scarce metals such as platinum, Lack of standardized manufacturing protocols.

Future Perspectives

Future research should focus on developing environmentally friendly and low-cost nanomaterials with improved efficiency and durability. Green synthesis techniques, recyclable nanomaterials, artificial intelligence-assisted material design, and hybrid nanocomposites are expected to play a significant role in next-generation energy technologies. Continued advances in nanotechnology will contribute to more efficient solar cells, safer batteries,

high-performance supercapacitors, and economically viable hydrogen production systems, supporting global efforts toward clean energy and sustainable development.

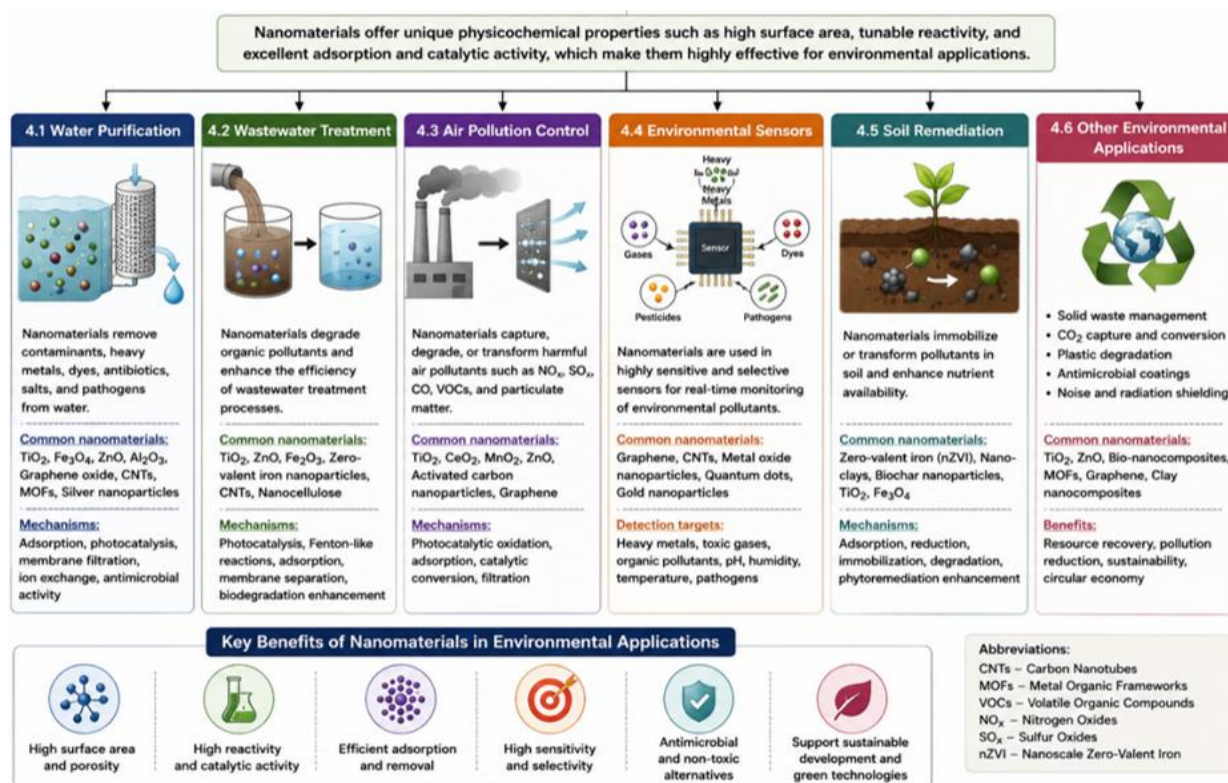


Fig. 2. Applications of Nanomaterials

Challenges in the Application of Nanomaterials

Although nanomaterials have demonstrated remarkable potential in energy generation, energy storage, and environmental remediation, several technical, economic, and regulatory challenges continue to hinder their large-scale commercialization. Addressing these challenges is essential to ensure the safe, sustainable, and widespread adoption of nanotechnology.

Toxicity and Human Health Concerns

The extremely small size of nanomaterials allows them to interact with biological systems in ways that are not yet fully understood. Some nanoparticles may penetrate cell membranes, accumulate in tissues, and induce oxidative stress, inflammation, or cytotoxicity. Therefore, comprehensive toxicological studies are necessary before large-scale industrial and biomedical applications.

Environmental Risks

Nanoparticles released during manufacturing, transportation, usage, or disposal may enter soil, water bodies, and the atmosphere. Their long-term ecological impacts on microorganisms, aquatic organisms, plants, and wildlife remain uncertain. Sustainable disposal and recycling strategies are therefore required. High Production Costs: Many high-performance nanomaterials require sophisticated synthesis techniques, expensive precursor materials, and advanced characterization equipment. These factors increase production costs and limit commercial-scale implementation. Large-Scale Manufacturing: Producing nanomaterials with consistent particle size, morphology, purity, and reproducibility remains a significant challenge. Industrial-scale production requires standardized synthesis methods that are both economical and environmentally friendly. Stability and Durability: Certain nanomaterials undergo oxidation, aggregation, or structural degradation during long-term operation, reducing their efficiency in solar cells, batteries, catalysts, and environmental treatment systems. Regulatory and Ethical Issues: International regulations governing the production, transportation, commercialization, and disposal of nanomaterials are still evolving. Clear safety guidelines, standardized testing protocols, and life-cycle assessments are essential for responsible innovation.

The future of nanotechnology is closely linked to the development of sustainable energy systems and environmental conservation. Continued research is expected to overcome current limitations through innovative materials and advanced manufacturing techniques. Green Nanotechnology: Green synthesis methods using plant extracts, microorganisms, and biodegradable materials are gaining attention because they reduce toxic chemicals, lower production costs, and minimize environmental impacts. Advanced Energy Storage: Next-generation nanomaterials are expected to improve lithium-ion batteries, sodium-ion batteries, solid-state batteries,

supercapacitors, and hydrogen storage systems by increasing energy density, charging speed, and operational safety. Artificial Intelligence-Assisted Material Design: Artificial intelligence (AI), machine learning, and computational materials science are accelerating the discovery of novel nanomaterials with optimized structural and functional properties for energy and environmental applications. Smart Environmental Monitoring: Nanotechnology-based sensors integrated with the Internet of Things (IoT) will enable real-time monitoring of air quality, water quality, industrial emissions, and environmental pollutants, supporting smart cities and sustainable infrastructure. Carbon Capture and Climate Change Mitigation: Nanostructured adsorbents and catalysts are expected to play a vital role in carbon dioxide capture, conversion, and storage technologies, contributing to global climate change mitigation efforts. Circular Economy: Future research will focus on recyclable, biodegradable, and reusable nanomaterials that promote resource efficiency, waste reduction, and sustainable industrial development.

Conclusion

Nanomaterials have emerged as one of the most transformative technologies for addressing global challenges related to sustainable energy production and environmental protection. Their unique physicochemical properties, including high surface area, superior catalytic activity, enhanced electrical conductivity, and tunable surface characteristics, have enabled significant improvements in renewable energy systems, energy storage devices, pollution control technologies, water purification, and environmental monitoring. In the energy sector, nanomaterials have enhanced the efficiency of solar cells, lithium-ion batteries, supercapacitors, fuel cells, and hydrogen production systems. In environmental applications, they have demonstrated excellent performance in wastewater treatment, air purification, heavy metal removal, photocatalytic degradation of pollutants, soil remediation, and environmental sensing. These advancements contribute directly to achieving global sustainability goals and reducing dependence on conventional fossil fuels. Despite these remarkable achievements, challenges related to toxicity, environmental safety, manufacturing costs, large-scale production, and regulatory frameworks must be carefully addressed to ensure the safe commercialization of nanotechnology. Future research should emphasize green synthesis methods, multifunctional nanomaterials, recyclable nanocomposites, and intelligent nanotechnology integrated with artificial intelligence and smart monitoring systems. Overall, nanomaterials are expected to play a central role in the transition toward clean energy, environmental sustainability, and a circular economy. Continued interdisciplinary collaboration among scientists, engineers, policymakers, and industry stakeholders will accelerate the development of safe, efficient, and economically viable nanotechnologies, helping to build a cleaner, healthier, and more sustainable future.

References

- Ahamed M, AL Salhi MS and Siddiqui MKJ (2010) Silver nanoparticle applications and human health. *Clinica Chimica Acta* 411(23–24):1841–1848. <https://doi.org/10.1016/j.cca.2010.08.016>
- Ahmed S, Khan M and Ali Z (2024) Nanotechnology in environmental monitoring and pollution control. *Environmental Research* 248:118079.
- Boisseau P and Loubaton B (2011) Nanomedicine, nanotechnology in medicine. *Comptes Rendus Physique* 12(7):620–636. <https://doi.org/10.1016/j.crhy.2011.06.001>
- Chen L, Zhao X and Wang J (2025) Emerging nanomaterials for environmental remediation and clean energy technologies. *Advanced Functional Materials* 35(4):2401234.
- Chen X and Mao SS (2007) Titanium dioxide nanomaterials: synthesis, properties, modifications, and applications. *Chemical Reviews* 107(7):2891–2959. <https://doi.org/10.1021/cro500535>
- Dresselhaus MS and Thomas IL (2001) Alternative energy technologies. *Nature* 414:332–337. <https://doi.org/10.1038/35104599>
- Ferrari M (2005) Cancer nanotechnology: opportunities and challenges. *Nature Reviews Cancer* 5(3):161–171. <https://doi.org/10.1038/nrc1566>
- Gao W (2015) *The chemistry of nanomaterials*. Elsevier.
- Geim AK and Novoselov KS (2007) The rise of graphene. *Nature Materials* 6(3):183–191. <https://doi.org/10.1038/nmat1849>
- Goodenough JB and Park KS (2013) The Li-ion rechargeable battery: a perspective. *Journal of the American Chemical Society* 135(4):1167–1176. <https://doi.org/10.1021/ja3091438>
- Gupta R, Patel S and Sharma D (2024) Nanomaterials for carbon capture and sustainable development. *Journal of Environmental Chemical Engineering* 12(2):112345.
- Iijima S (1991) Helical microtubules of graphitic carbon. *Nature* 354:56–58. <https://doi.org/10.1038/354056a0>
- Jariwala D, Sangwan VK, Lauhon LJ, et al. (2014) Emerging device applications for semiconducting two-dimensional transition metal dichalcogenides. *ACS Nano* 8(2):1102–1120.

- Kamat PV (2007) Meeting the clean energy demand: nanostructure architectures for solar energy conversion. *Journal of Physical Chemistry C* 111(7):2834–2860. <https://doi.org/10.1021/jp066952u>
- Kumar S and Kumar A (2023) Nanomaterials for water purification and environmental remediation. *Environmental Nanotechnology, Monitoring & Management* 19:100785.
- Li H, Wang X and Chen Y (2024) Nanomaterials for advanced energy storage systems: a review. *Journal of Energy Storage* 79:110235.
- Li Y, Xu G and Zhang C (2022) Nanomaterials for sustainable energy storage and conversion. *Nano Energy* 95:107024.
- Liu J, Qiao SZ, Hu QH, et al. (2011) Magnetic nanocomposites with mesoporous structures. *Small* 7(4):425–443.
- Novoselov KS, et al. (2004) Electric field effect in atomically thin carbon films. *Science* 306(5696):666–669. <https://doi.org/10.1126/science.1102896>
- Pan X, Yang MQ, Fu X, et al. (2013) Defective TiO₂ with oxygen vacancies. *Nanoscale* 5:3601–3614.
- Peer D, et al. (2007) Nanocarriers as an emerging platform for cancer therapy. *Nature Nanotechnology* 2(12):751–760.
- Qu X, Alvarez PJJ and Li Q (2013) Applications of nanotechnology in water treatment. *Water Research* 47(12):3931–3946.
- Rao CNR, Müller A and Cheetham AK (2004) The chemistry of nanomaterials. Wiley-VCH.
- Service RF (2018) New generation batteries powered by nanotechnology. *Science* 361(6403):642–644.
- Sharma P, Singh V and Gupta R (2023) Green nanotechnology for sustainable energy applications. *Renewable and Sustainable Energy Reviews* 176:113183.
- Shi J, Kantoff PW, Wooster R, et al. (2017) Cancer nanomedicine: progress, challenges and opportunities. *Nature Reviews Cancer* 17(1):20–37.
- Singh R and Lillard JW (2009) Nanoparticle-based targeted drug delivery. *Experimental and Molecular Pathology* 86(3):215–223.
- Slater MD, Kim D, Lee E, et al. (2013) Sodium-ion batteries. *Advanced Functional Materials* 23(8):947–958.
- Sun Y and Xia Y (2002) Shape-controlled synthesis of gold and silver nanoparticles. *Science* 298(5601):2176–2179.
- Verma P, Singh A and Kumar R (2025) Recent advances in nanomaterials for renewable energy applications. *Nano Energy* 120:109873.
- Wang H and Dai H (2013) Strongly coupled inorganic–nano-carbon hybrid materials. *Chemical Society Reviews* 42:3088–3113.
- Wang Q and Domen K (2020) Particulate photocatalysts for water splitting. *Chemical Reviews* 120(2):919–985.
- Whitesides GM (2005) Nanoscience, nanotechnology, and chemistry. *Small* 1(2):172–179.
- Xia Y, Yang P, Sun Y, et al. (2003) One-dimensional nanostructures. *Advanced Materials* 15(5):353–389.
- Xu M, Liang T, Shi M, et al. (2013) Graphene-like two-dimensional materials. *Chemical Reviews* 113(5):3766–3798.
- Yang Z, Zhang J, Kintner-Meyer M, et al. (2011) Electrochemical energy storage. *Chemical Reviews* 111(5):3577–3613.
- Zhang L, Gu FX, Chan JM, et al. (2008) Nanoparticles in medicine. *Clinical Pharmacology & Therapeutics* 83(5):761–769.
- Zhang X, et al. (2021) Advanced nanomaterials for renewable energy conversion. *Nano Energy* 85:105974.
- Zhao Y, et al. (2022) Nanotechnology for environmental sustainability. *Journal of Cleaner Production* 350:131438.
- Zhu Y, Murali S, Cai W, et al. (2010) Graphene and graphene oxide. *Advanced Materials* 22(35):3906–3924.

Author Contributions

VG conceived the concept, wrote and approved the manuscript.

Acknowledgements

Not applicable.

Funding

Not applicable.

Availability of data and materials

Not applicable.

Competing interest

The authors declare no competing interests.

Ethics approval

Not applicable.



Open Access *This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution, and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third-party material in this article are included in the article's Creative Commons license unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. Visit for more details <http://creativecommons.org/licenses/by/4.0/>.*

**The authors are solely responsible for the accuracy, integrity and content of this paper. The views, opinions, findings and conclusions expressed herein are those of the authors and do not necessarily reflect the views or policies of the publisher, editors, reviewers or affiliated institutions.*

Citation: Geetha V (2026) Nanomaterials and their Applications in Energy and Environment: A Comprehensive Review. *Environmental Science Archives* 5(2): 449-455.